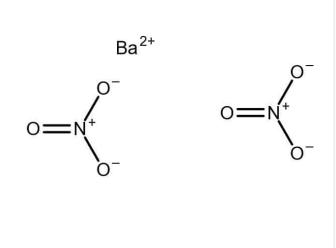


# h1>Barium nitrate 99.5% class EL-[10022-31-81



CAS number: **10022-31-8**Summary formula: **Ba (NO3) 2**Molar mass: **261.32 g / mol** 

Synonyms: none

Translation [ENG]: barium nitrate
Application: Barium nitrate is used in
industry for the production of green
signal lamps, for removing gases from
vacuum lamps and for the production of
barium oxide. Barium nitrate burns with
a bright green flame and is used in
signal flares and pyrotechnics.

#### **VARIATIONS**

| Image                    |                   | Price                         | Pack size |
|--------------------------|-------------------|-------------------------------|-----------|
| Ba <sup>2+</sup> O=N+ O- | 0=\(\frac{0}{0}\) | £132,96 gross   £108,10 netto | 10 kg     |



| Image                           | Price                         | <b>Pack size</b><br>25 kg |
|---------------------------------|-------------------------------|---------------------------|
| $0 = N_{+}^{0}$ $0 = N_{+}^{0}$ | £303,81 gross   £247,00 netto |                           |

#### PRODUCT DESCRIPTION

### Barium nitrate 99.5% class EL [10022-31-8]

Barium nitrate is used in industry for the production of green signal lamps, for removing gases from vacuum lamps and for the production of barium oxide. Barium nitrate burns with a bright green flame and is used in signal flares and pyrotechnics. It can be produced by treating barium carbonate with nitric acid.

Density: 3.2 g / cm3 (20 ° C)
Melting point: 592 ° C

PH value: 5.0 - 8.0 (50 g / l, H<sub>2</sub>O, 25 ° C)

Bulk density: 1900 g / m3 Solubility: 94 g / l

PH value (5%; water; 25 ° C): 5.0 - 8.0

Chloride (Cl): ≤ 0.0005%

Heavy metals (as Pb):  $\leq 0.0005\%$ 

Ca (calcium):  $\leq 0.002\%$ Fe (iron):  $\leq 0.0002\%$ 

K (potassium):  $\leq 0.005\%$ Na (sodium):  $\leq 0.005\%$ Sr (strontium):  $\leq 0.05\%$ 

## **Hazard pictograms**

Labeling of hazardous chemicals and mixtures that are part of the Globally Harmonized System of Classification and Labeling of Chemicals (GHS). The pictograms recommended by GHS have the



shape of a square set on the top. They should contain a black symbol on a white background with a red border.

Priority rules to be observed in connection with the labeling of a substance:

- the skull and crossbones, the exclamation mark pictogram should not be added.
- corrosive effect, the exclamation mark pictogram should not be added if it concerns eye or skin irritation.
- health hazard determining respiratory sensitization, the exclamation mark pictogram should not be added if it concerns skin sensitization or irritation to eyes or skin.

Source: **GHS pictograms**